Name: \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ Period: \_\_\_\_

Review for Ionic Compounds Nomenclature:

Name the following compounds containing polyatomic ions:

1. Na2SO4
2. K3PO4
3. Fe(NO3)3
4. Mn(OH)2
5. Na2SO3

Name the following acids:

1. HF (aq)
2. HNO3 (aq)
3. HNO2 (aq)
4. HBr (aq)
5. HI (aq)
6. H2SO4 (aq)
7. H2SO3 (aq)
8. HC2H3O2 (aq)
9. H2S (aq)
10. HClO2 (aq)

Name the following compounds:

1. KOH

2. Na2CO3

3. HNO3 (aq)

4. Co(NO3)3

5. CaCl2

6. PbO2

7. PbO

8. NH4ClO4

9. FeBr2

10. HgCl2

Give the formula for the following compounds

1. Sodium Iodide
2. Tin (II) Flouride
3. Potassium Sulfide
4. Calcium Iodide
5. Chromium (III) nitrite
6. Mercury (II) Bromide
7. Acetic Acid
8. Calcium Hydrogen Carbonate
9. Magnesium Sulfide
10. Barium Hydroxide
11. Strontium Phosphate
12. Hydrosulfuric Acid
13. Sulfuric Acid
14. Sulfurous Acid
15. HydroBromic Acid
16. Hydrocyanic acid
17. Nitrous acid
18. Nitric acid
19. Lead (IV) Oxide
20. Lead (II) Oxide